## TOTAL STEREOSPECIFIC SYNTHESIS OF (+) AZIMIC AND (+) CARPAMIC ACIDS FROM D-GLUCOSE

Stephen Hanessian and Richard Frenette

Department of Chemistry, University of Montreal Montreal, Quebec, H3C 3Vl

Summary: The total synthesis of (+)-azimic and (+)-carpamic acids are described, based on a biomimetic heterocyclization of optically active precursors derived from D-glucose,

The piperidine alkaloids constitute a large family of naturally occurring substances many of which possess important physiological and antibiotic properties<sup>2</sup>. The macrocyclic dilactones azimine<sup>3</sup> and carpaine<sup>4</sup> belong to a small subgroup containing a 2,3,6-trisubstituted piperidine skeleton whose constitutional structures and stereochemistry have been arrived at by spectroscopic and degradative studies. Azimic 1 and carpamic 2 acids, are stereochemically related and have been shown to be  $2(S)$ -methyl-3(S)-hydroxy-6(R)-(carboxyalkyl)piperidines with an "all cis" configuration, a stereochemical feature found in other piperidinoltype alkaloids as well<sup>5</sup>. We describe herein a total and stereospecific synthesis of  $\downarrow$  and  $\downarrow$ from D-glucose, based on the concept of "chiral templates"<sup>6</sup>, as illustrated in the retrosynthetic analysis shown below.



The synthesis of (+)-carpamic acid starts with methyl a-D-glucopyranoside 2 which can be transformed into the unsaturated derivative  $\cancel{z}$  in high overall yield by known procedures'.

 $12$ 









 $_{\text{11}}$ 





i. 5% Pd/C, H<sub>2</sub>, MeOH (quant.); ii. NBS/CCl<sub>4</sub>, reflux, 30 min. (95%); iii. NaOMe, MeOH; iv. LAH. THF, 25°; v. BnBr. NaH, DMF (86% overall); vi. EtSH, HCl (71%); vii. TsCl, pyr. (98%); viii. NaN<sub>3</sub>, DMF, 80°, o.n (98%); ix. Br<sub>2</sub>, ether, H<sub>2</sub>O (63%); x. BrMg(CH<sub>2</sub>)<sub>7</sub>CH<sub>2</sub>OTHP, THF, -50°, 30 min. (48-53%); 80% based on recovered 10); xi. pyridinium chlorochromate NaOAc, CH<sub>2</sub>C1<sub>2</sub>, o.n. (83%); xii. Pd/C, H<sub>2</sub>, EtOAc, 85%; xiii. CbzCl, aq. acetone (88%); xiv. CrO<sub>3</sub>, aq. H<sub>2</sub>SO<sub>4</sub>, acetone, (57%), xv. 10% Pd/C,  $H_2$ , MeOH, 8h (83%).

3393

Selective reduction of the double bond, followed by treatment with NBS gave  $\mathfrak z$  as a syrup  $^9,$ [ $\alpha$ ]<sub>n</sub> + 130°. Sequential reduction, <u>0</u>-benzylation to 6, [ $\alpha$ ]<sub>n</sub> + 141.2°, followed by ethanethiolysis gave  $\zeta$  (syrup). The tosylate  $\zeta$ ,  $\alpha$   $I_n + 3.8$ <sup>o</sup>, prepared in the usual way, was then treated with excess sodium azide in hot DMF to give the desired azido derivative 2 as a syrup  $[\alpha]_p + 9.1^\circ$ . Among several procedures attempted to regenerate the aldehyde function from  $\zeta$ , none were as effective as that utilizing bromine in aqueous ether $^{10}$ , which led to the syrupy aldehydo derivative  $Q_{0}$ ,  $\alpha$   $\beta_{p}$  - 9.45°. Completion of the synthesis resolved itself in the introduction of the acid side chain, and the reductive cyclization<sup>11</sup> of an aminoketone derivative generated from 12 as called for in our synthetic plan. A high degree of stereocontrol was expected in this step, since the intermediate  $\Delta^1$ -piperideine derivative should be hydrogenated $^{12}$  from the  $\alpha$ -face of the molecule. Treatment of the aldehyde  $\gimel \mathfrak{g}$  with the Grignard reagent $^{13}$  prepared from 8-bromo-l(tetrahydropyran-2-yloxy)octane led to the formation of  $\downarrow\downarrow$ , presumably as a mixture of epimers. Oxidation with pyridinium chlorochromate  $^{14}$  produced the azidoketone derivative  $12$ ,  $\lceil \alpha \rceil$  - 7.3<sup>°</sup>, which upon hydrogenation, gave 13. Conversion to the N-benzyloxycarbonyl derivative  $\begin{bmatrix} \alpha & 0 \end{bmatrix}$  - 18.7°, followed by oxidation of the terminally protected primary alcohol led directly to N-benzyloxycarbony1-3-0-benzyl carpamic acid  $\downarrow\!\phi$ ,  $[\alpha]_{\text{n}}$  - 22.8°, m/e 338 (M<sup>+</sup>- C<sub>13</sub>H<sub>25</sub>O<sub>2</sub>). Finally, reductive removal of the N,O-protecting groups gave crystalline carpamic acid 2 mp 225-227° (EtOH, acetone<sup>15</sup>); [a ]<sub>n</sub> + 5°; M<sup>+</sup>calcd. 257.2021; meas. 257.1990, lit.<sup>16</sup> mp 225-226; [a ]<sub>D</sub> +  $7^{\circ}$ <sup>17</sup>. The transformation of carpamic acid into carpaine has been reported by Corey and coworkers  $^{18}$ . Following the same sequence, but using the Grignard reagent $^{19}$  derived from 6-bromo-1-(tetrahydropyran-2-yloxy)hexane, it was possible to prepare crystalline azimic acid in the same relative yields as described for carpamic acid. Azimic acid showed mp 214-215<sup>°</sup> (EtOH)<sup>15</sup>; [a]<sub>n</sub> + 8<sup>°</sup> (MeOH): M<sup>+</sup> calcd. 229. 1677; meas. 229.1620, and was identical (t.l.c., mass spectrum) with a sample of ( $\pm$ )-azimic acid <sup>20</sup>.

In addition to being stereospecific, the synthetic approach to piperidinol alkaloids shown in this paper, is also a biomimetic one $^{21}$  and of general applicability to other, more complex alkaloidal structures, as well as related heterocycles $^{22}.$ 

In an alternate approach, the aldehyde 10 was condensed with carboethoxymethylene triphenylphosphorane (CH<sub>2</sub>C1<sub>2</sub>, 25<sup>°</sup>, 2h) to give the corresponding olefinic derivative as a syrup (57%). Treatment with Ra-Ni in ethanol, followed by N-acetylation led directly to the desired the N-acetylpiperidine derivative,  $(m/e 290, M<sup>-</sup> - Ac)$ , presumably as a mixture of epimers at C-6, which could be debenzylated to the corresponding hydroxypiperidines (m/e 220,  $M^+$ - Ac; m/e 156, N-acetyl 2-methyl-3-hydroxy- $\Delta^1$ -piperideinium ion).



The stereochemistry of the intramolecular addition and further chemical modifications in this series are presently under study. This approach constitutes another type of biomimetic heterocycliaation 23 , involving precursors derived from optically active, acyclic carbohydrates and provides an alternate strategy for the asymmetric synthesis of piperidine and pyrrolidine alkaloids.

Acknowledgement: We thank the National Research Council of Canada and the Ministère de 1'Education du Qu6bec for a fellowship to R.R. and for financial assistance. We also thank Dr. E. Brown for a sample of ( $\pm$ ) azimic acid, R. Mayer for recording 90MHz spectra, and Carol Lepine and F. Messier for the mass spectra.

## References and Notes:

- 1. D. Gross, Progress Chem. Org. Nat. Prod., 29, 1 (1971); R.K. Hill in the Alkaloids S.W. Pelletier, ed., p. 395, Van Nostrand Reinhold, New York, 1970.
- 2. See for example, G. Fodor, J.-P. Fumeaux and V. Sankaran, Synthesis 464 (1972).
- 3. T.M. Smallberger, G.J.H. Rall and H.L. de Waal, Tetrahedron, 24, 6417 (1968); and references cited therein.
- 4. J.L. Coke and W.Y. Rice, Jr., J. Org. Chem., 30, 3420 (1965); and references cited therein.
- 5. See for example (+)-spectaline and (+)-prosopsine, which have the same  $(2S)$ ,  $3(S)$ ,  $6(R)$ configuration;  $(-)$ -cassine is  $(2R)$ ,  $3(R)$ ,  $6(S)$ ; ref. 1.
- 6.
- 7. S. Hanessian, <u>Acc. Chem. Res.</u>, 12, 159 (1979), and references cited therein.<br>D. Horton, J.K. Thompson and C.G. Tindall, Jr., <u>Methods Carbohyd. Chem</u>., <u>6</u>, 297 (197**2**) S. Hanessian, A. Bargiotti and M. LaRue, Tetrahedron Lett., 737 (1978).
- 8. S. Hanessian, Methods in Carbohyd. Chem., 6, 183 (1972); S. Hanessian and N. R. Plessas J. Org. Chem., 34, 1035, 1045, 1053 (1969).
- 9. All compounds were characterized by spectroscopic and mass spectral methods.
- 10. J. Defaye, Bull.Chem. Soc. France, 2686 (1964).
- 11. S. Hanessian, J. Org. Chem., 34, 675 (1969); Chem., Ind. (London), 2126 (1966); S. Hanessian and T.H. Haskell, J. Heterocyclyc Chem., 1, 55 (1964).
- 12. R.K. Hill and T. Yuri, Tetrahedron, 33, 1569 (1977); see also A. Astier and M.M. Plat Tetrahedron Lett., 2051 (1978)
- 13. O.L. Chapman, K.C. Mattes, R.S. Sheridan and J.A. Klun, J. Am. Chem. Sot. 100, 4878 (1978).
- 14. E.J. Corey and J.W. Suggs, Tetrahedron Lett., 2647 (1975).
- 15. Recorded on a Reichert apparatus. At ca. 185-190°C the crystals sublime to give beautiful long needles which melt at the temperature range indicated; see also ref. 17.
- 16. H. Rapoport and H.D. Buldridge, Jr., J. Am. Chem. Soc.,  $74$ , 5365 (1952).
- 17. G. Barger, J. Chem. Soc., 466 (1910).
- 18. E.J. Corey, K.C. Nicolaou and L.S. Melvin, J. Am. Chem. Soc., 97, 654 (1975).
- 19. K. Mori, Tetrahedron, 30, 3807 (1974).
- 20. E. Brown and R. Dhal, Tetrahedron Lett., 1029 (1974).
- 21. For the biogenesis of piperidlne alkaloids, see E. Leete, J.C. Lechleiter and R.A. Carver, Tetrahedron Lett., 3779 (1975); E. Leete and R.A. Carver, J. Org. Chem., 40 2151 (1975); E. Leistner and R.D. Spencer, J. Am. Chem. Soc. 95 4715 (1973).
- 22. See for example J.P.H. Verheyden, A.C. Richardson, R.S. Bhatt, B.D. Grant, W.L. Fitch and J.G. Moffatt, Pure Appl. Chem., 50, 1363 (1978); T. Ogawa, T. Kawano and M. Matsui Carbohydr. Res., C 31 (1977); H. Ohrui and S. Emoto, Tetrahedron Lett., 2765 (1975); C.C. Dean and T.D. Inch, Chem. Commun., 813 (1969).
- 23. See for example J.J.J. de Boer and W.N. Speckamp, Tetrahedron Lett., 4039 (1975); T. Wakabayashi, K. Watanabe, Y. Kato and M. Saito, Chem.Lett., 223 (1977).

(Received in USA 1 May 1979)